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International Application No.	
International Filing Date	
Name of receiving Office and "PCT International Application"	,

REQUEST	
	International Filing Date
The undersigned requests that the present international application be processed according to the Patent Cooperation Treaty.	Name of receiving Office and "PCT International Application" Applicant's or agent's file reference
	(if desired) (12 characters maximum) \(\beta\) 2132-1 WO
Box No. I TITLE OF INVENTION	the state of the s
METHOD TO OBTAIN MICROPARTICLES	
Box No. II APPLICANT	
Name and address: (Family name followed by given name; for a designation. The address must include postal code and name of con address indicated in this Box is the applicant's State (that is, country of residence is indicated below.)	untry. The country of the This person is also inventor
AstraZeneca AB S-151 85 Södertälje	Telephone No. +46 8 553 260 00
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•	Teleprinter No.
State (that is, country) of nationality:	State (that is, country) of residence:
	the United States the States indicated in the States of America only the Supplemental Box
Box No. III FURTHER APPLICANT(S) AND/OR (FURT	HER) INVENTOR(S)
Name and address: (Family name followed by given name; for a designation. The address must include postal code and name of con address indicated in this Box is the applicant's State (that is, country of residence is indicated below.) SJÖBLOM, Brita AstraZeneca R&D Mölndal S-431 83 Mölndal Sweden	This person is: applicant only applicant and inventor inventor only lf this check-box is marked, do not fill in below.
State (that is, country) of nationality: SE	State (that is, country) of residence: SE
This person is applicant all designated all designate for the purposes of:	the States except that the United States the States indicated in the Supplemental Box
Further applicants and/or (further) inventors are indicated of	on a continuation sheet.
Box No. IV AGENT OR COMMON REPRESENTATIVE	; OR ADDRESS FOR CORRESPONDENCE
The person identified below is hereby/has been appointed to act of the applicant(s) before the competent International Authorities	on behalf agent common representative
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The fol	llowing designations are hereby made under Rule 4.9(a) (a)	mark	the ap	oplicable check-boxes; at least one must be marked):
	nal Patent			•
-	ARIPO Patent: GH Ghana, GM Gambia, KE Kenya, LS	S Les ganda	otho, a, ZW	MW Malawi, MZ Mozambique, SD Sudan, SL Sierra Leone, Zimbabwe, and any other State which is a Contracting State
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☑ EP	European Patent: AT Austria, BE Belgium, CH a DK Denmark, ES Spain, FI Finland, FR France, GB U	Unite	ed Kin	witzerland and Liechtenstein, CY Cyprus, DE Germany, agdom, GR Greece, IE Ireland, IT Italy, LU Luxembourg, ther State which is a Contracting State of the European Patent
☑ OA	OAPI Patent: BF Burkina Faso, BJ Benin, CF Centro GA Gabon, GN Guinea, GW Guinea-Bissau, ML Mali, other State which is a member State of OAPI and a Contra	MR acting	. Maur g State	n Republic, CG Congo, CI Côte d'Ivoire. CM Cameroon, itania, NE Niger, SN Senegal, TD Chad, TG Togo. and any e of the PCT (if other kind of protection or treatment desired.
Nation	al Patent (if other kind of protection or treatment desired, spe	cify (on dott	ed line):
_	United Arab Emirates	-		Saint Lucia
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				le above, the applicant also makes under Rule 4.9(b) all other
				on(s) indicated in the Supplemental Box as being excluded onal designations are subject to confirmation and that any
				e priority date is to be regarded as withdrawn by the applicant
at the ex	xpiration of that time limit. (Confirmation (including fees) mus	st rea	ich the	receiving Office within the 15-month time limit.)

Sheet No. .3....

Box No. VI PRIORITY C	LAIM	Further pr	iority claims are indicated	I in the Supplemental Box.	
Filing date	Number		Where earlier application is:		
of earlier application (day month/year)	of earlier applicati	national application: country	regional application:* regional Office	international application: receiving Office	
(10.09.1999) 10 September 1999	9903236-9	Sweden			
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Where the earlier application is Convention for the Protection of In	an ARIPO application, i	t is mandatory to indicate in the S sich that earlier application was fi	Supplemental Box at least of the Children Country Supplemental Box at least of the Children Coun	ne country party to the Paris	
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Choice of International Search (1) two or more International Search competent to carry out the intern	arching Authorities are	Request to use results of essearch has been carried out by	arlier search; reference or requested from the Interna	to that search (if an earlier tional Searching Authority):	
the Authority chosen; the two-letter		Date (day/month/year)	Number	Country (or regional Office)	
ISA/ SE		25 April 2000	SE 99/01185	Sweden	
Box No. VIII CHECK LIST					
This international application of the following number of sheet	ts:	ational application is accompa	nnied by the item(s) mark	ed below:	
request : 3	-	calculation sheet arate signed power of attorney			
description (excluding sequence listing part) : 15	- '	y of general power of attorney		v. GF 2141/ 2000	
claims : 3	l <u> </u>	ement explaining lack of signa		.y. 0. 2	
abstract : 1	5. 🔲 prio	rity document(s) identified in	Box No. VI as item(s):		
drawings :	6. 🔲 tran	slation of international applica	ation into (language):		
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should accompany the abstract		international application:	English		
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PATENT COOPERATION TREATY

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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

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Applicant's or a	igent's file reference	COD CUDTUED A	27:21		ation of Transmittal of International
H 2132-1 W	O	FOR FURTHER A	CHON	Preliminary	Examination Report (Form PCT/IPEA/416)
International ap	oplication No.	International filing date	(day/month	/year)	Priority date (day/month/year)
PCT/SE00/0)1682	01/09/2000			10/09/1999
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Applicant					
ASTRAZEN	ECA AB et al.			i	
This inter and is tra	national preliminary exami nsmitted to the applicant a	nation report has beer ccording to Article 36.	prepared	by this Inter	mational Preliminary Examining Authority
2. This REP	ORT consists of a total of	5 sheets, including thi	is cover sh	eet.	
been	report is also accompanied amended and are the bas Rule 70.16 and Section 60	is for this report and/or	r sheets co	ntaining rec	, claims and/or drawings which have tifications made before this Authority e PCT).
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3. This report	t contains indications relat	ing to the following iter	ms:		
ı 🛭	Basis of the report				
H 🗀	·				•
III 🗆	Non-establishment of op	inion with regard to no	ovelty, inve	entive step a	nd industrial applicability
IV 🗆			-	•	., .
v 🛭	Reasoned statement und citations and explanation	der Article 35(2) with rons suporting such state	egard to no	ovelty, inven	tive step or industrial applicability;
vi 🗆	Certain documents cited	t			
VII 🛛	Certain defects in the int	ernational application			1
VIII 🗆	Certain observations on	the international applic	cation		
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Date of submissi	on of the demand		Date of co	mpletion of th	is report
26/03/2001	•			2 2 08	R. 01
Name and mailin	g address of the international		Authorized	officer	(500 ay)

preliminary examining authority:

European Patent Office
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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/SE00/01682

I.	Bas	sis of the report					,
1.	the and	receiving Office in re	ents of the international sponse to an invitation this report since they do	under Article	14 are referred to	in this report as	"originally filed"
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2.	Witl	h regard to the langu guage in which the int	age, all the elements meternational	narked above was filed, unle	were available or ss otherwise indi	furnished to this cated under this i	Authority in the tem.
	The	se elements were av	ailable or furnished to t	his Authority in	n the following lar	nguage: , which	is:
		the language of a tra	anslation furnished for t	he purposes o	of the internationa	al search (under F	Rule 23.1(b)).
		the language of pub	lication of the internation	nal application	n (under Rule 48.	.3(b)).	
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3.			eotide and/or amino ad examination was carrie				ication, the
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		filed together with th	e international applicat	ion in compute	er readable form.		
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			he subsequently furnishication as filed has be		quence listing do	es not go beyond	I the disclosure in
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4.	The	amendments have r	esulted in the cancellat	ion of:			
		the description,	pages:			·	
		the claims,	Nos.:	•			
		the drawings,	sheets:				
5.		This report has beer	n established as if (som	e of) the ame	ndments had not	been made, sinc	e they have been

considered to go beyond the disclosure as filed (Rule 70.2(c)):

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/SE00/01682

(Any replacement sheet containing such amendments must be referred to under item 1 and annexed to this report.)

6. Additional observations, if necessary:

V. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N)

Yes:

Claims 1-20

No:

Claims

Inventive step (IS)

Yes:

Claims 1-20

No: Claims

Industrial applicability (IA)

Yes:

Claims 1-20

No: Claims

2. Citations and explanations see separate sheet

VII. Certain defects in the international application

The following defects in the form or contents of the international application have been noted: see separate sheet

EXAMINATION REPORT - SEPARATE SHEET

Re Item V

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Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

Reference is made to the following documents: 1.

D1: GB-A-2329124 D2: US-A-5017383 D3: US-A-5 102 983

D3, which is cited in the description, was not cited in the International Search Report. A copy of the document is appended hereto.

Prior art document D1 discloses a method of producing dose units of active 2. agents comprising (i) the spray-freezing of a solution or suspension of an active agent and a polymer to obtain frozen particles and (ii) subsequent freeze drying (p. 2, l. 21 - p. 3, l. 2 and p. 4, l. 14-17).

D1 differs from the subject-matter of the present application in that no percentages of the components are indicated and thus neither the dry content nor the content of polymer of the composition.

A method of producing a fine coated pharmaceutical preparation comprising (i) spray-freezing of a solution of a drug and a binder, (ii) classifying, (iii) heating and (iv) coating of these particles and (v) subsequent drying in a fluidized bed is disclosed in D2 (ex. 1). The dry content of this composition is 33 % and the amount of polymer based upon the dry content of the medium 6 %.

In contrast to the subject-matter of the present application the particles are heated to melt the liquid medium before coating and the coated particles are dried on a fluidized bed (D2: Col. 4, I. 24-25 and col. 5, I. 3-5) whereas dry particles are obtained in the present application by sublimating the frozen liquid.

D3 discloses a process for preparing foamed bioabsorbable polymer particles for surgical use, which is identical to the one of the present application. Apart from the fact that there are no examples comprising apart from the polymer a pharmaceutically active substance, the porosity and the pore sizes of the particles ob-

EXAMINATION REPORT - SEPARATE SHEET

tained in D3 are very large (claims 1, 12 and col. 5, l. 11-13).

In view of the cited prior art the subject-matter of the present application seems to be new and fulfil the requirements of Art. 33(2) PCT.

The problem of the present application seems to be the provision of an improved method of preparing homogeneous microparticles with a low friability and a high 3. content of active substance (p. 5, l. 17-27).

Although the claimed production steps of spray-freezing and freeze-drying for preparing microparticles are known as well as the fact that the polymer concentration influences the density of the formed particles (D3: col. 2, I. 66-68) there is no indication in the cited prior art that a minimum dry content of 15 vol. % and a polymer content of at least 5 % based on the dry content solves the problem of the present application.

Therefore, the subject-matter of claims 1-20 seems to fulfil the requirements of Art. 33(3) PCT.

The requirements of Rule 5.1(a)(ii) PCT are fulfilled as the prior art is discussed 4. on pages 2-5 of the description.

Re Item VII

Certain defects in the international application

- For clarity reasons the term "other substances" (claims 1 and 13) should be substituted by the definition given on p. 10, l. 1-4 of the description (Art. 6 PCT). 1.
- The expressions such as "optionally and preferably" employed in claims 1 and 10 relate to preferred modes which should be claimed in dependent claims. 2.
- As the expression "not limited to" is not deleted in the description (p. 8, 10, 11) the 3. requirements of Rules 5 and 9.1 PCT are not met.

(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 22 March 2001 (22.03.2001)

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- (72) Inventor; and
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Published:

- With international search report.
- Before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments.

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

19345

(54) Title: METHOD TO OBTAIN MICROPARTICLES

(57) Abstract: Method of obtaining microparticles. A method for the preparation of homogeneous microparticles containing a pharmaceutically active substance by a spray freezing technique wherein the medium to be atomized into droplets has a high dry content and comprises besides the active substance a polymer and a liquid (in which the polymer may be soluble) in which the active substance and polymer are suspended, dissolved or emulsified.

WO 01/19345 PCT/SE00/01682

METHOD TO OBTAIN MICROPARTICLES

Field of invention

The present invention provides a method of obtaining microparticles by a spray freezing technique. More specifically the present invention relates to a method by which spherical microparticles containing one or more pharmaceutically active substances can be prepared.

Background of the invention

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The strategy for pharmaceutical formulation of a given drug depends on different factors. Ultimately, these factors emanate from 1) the therapeutic needs, 2) the physical chemical properties of the drug, and 3) the influence of the biological environment where the formulation will release its contents. Thus, both technical and biopharmaceutical considerations will contribute to a successful therapy.

However, improved drug administration may also be achieved by so called modified release of the drug, which has been discussed extensively in the literature, e g R L Langer and D L Wise (Eds) "Medical Applications of Controlled Release", vols I, II (1984), CRC Press Inc, Boca Raton.

Several approaches to achieve different types of modified release are described in the references above. Of special importance to the present invention is modified release achieved by formulating the active substance with a suitable carrier material in the form of microparticles. Such a formulation then contains multiparticulate discrete delivery units, each of which can be coated if necessary with, e g a suitable pH sensitive, semipermeable or other polymeric film. Several advantages can be obtained with this type of formulation compared with more conventional delivery means. Thus, the small size of the

microparticles assures a fast and predictable emptying from the stornach, which is of special importance in the presence of food. Further, the particles will spread over a larger area in the whole GI-tract compared with a conventional monolithic (single-unit) formulation. This will result in a safer therapy when the active substance has local irritating side effects. Controllable plasma levels of absorbed drug can also be obtained. The microparticle formulation will also have a longer residence time in the colon which makes 24 hrs extended release formulations possible. From a technological point of view, microparticles are more suitable for coating and handling since a technical fault during the process may be serious for single unit formulations but less so for micropellets. Also, microparticle formulations are more easily manufactured and prepared in different doses than standard tablet systems.

Prior Art

An ideal method for the preparation of microparticles where the drug is homogeneously distributed within a polymeric matrix, should be simple, reproducible, rapid and minimally dependent on the solubility characteristics of the drug. A high product yield and a high degree of retention of the active substance in the final microparticles should also be obtained.

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Several different techniques are available for making microparticles (< 1 mm), e g spraydrying, extrusion-spheronization, spray-chilling, emulsion solvent evaporation/extraction and coating of nonpareil spheres, among others. A recent review was presented by Conti et al in STP Pharma Sci 7, 331 (1997) where the technical aspects of coacervation, spraydrying, emulsion solvent extraction, and emulsion solvent evaporation were discussed.

However, all existing techniques suffer from one or more drawbacks. Thus, many drugs are sensitive to heat and therefore will deteroriate which restricts the use of spray-drying or spray-chilling.

In extrusion spheronization and in coating of non-pareils particles it has been difficult to achieve acceptable microparticles in the size range of 50 - 400 µm. Pellets made by these methods contain significant amounts of inert excipients. This may make the pelletization of high-dose drugs by these methods a difficult task.

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Finally, in emulsification solvent evaporation, an emulsion has to be made and the drug to be incorporated is preferably lipophilic, which restricts the drugs which can be used.

Another drawback is the toxicity of the solvent used, usually methylene chloride, which can remain in the microparticles after drying.

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However, despite the many different approaches there has not been disclosed a technique that can produce both smaller microparticles but also particles of more uniform size. It is important to avoid, e g segregation and dose variation during further processing into capsules or tablets. Further, the existing techniques do not incorporate several desirable aspects such as the possibility to produce spherical microparticles of different size ranges that are homogeneous, have a high drug content and sufficient mechanical strength (to e g withstand coating processes) into one single technique.

A spray-freezing technique has been used for the processing and granulation of ceramic materials to achieve homogeneous distribution of additives within granules to be compacted. For the processing of slurries containing silicon-nitride, sintering additives and a binder, spherical free-flowing granules have been prepared by spray-freezing and subsequent freeze-drying. The homogenity of the slurry was retained in the granules and

thus in the final sintered product (Nyberg et al, Euro-Ceramics II 1, 447 (1993)).

Suspensions of silicon carbide and additives were processed in this way to give granules for compaction (US Patent 4,526,734). The increased homogenity compared with

traditional granulation techniques resulted in better mechanical properties of a whisker reinforced ceramic (EP 0 584 051). The process is also feasible for making homogeneous

powder blends for ceramic superconductors (Japanese unexamined patent application no.

30 59-102433).

Normally pharmaceutical materials are lyophilized by freeze-drying in a bulk process in which the solution/suspension to be freezed is placed in vials or on trays in a freeze-drier, where freezing and subsequent sublimation of the dry solvent take place. The dried product is a powder cake.

The rapid freezing provided by spray-freezing ensures that no concentration gradients exist in the resulting frozen particles and degradation of biological material is prevented. This approach has been used to achieve precise metering and dispensing (M J Akers and D J Schmidt, BioPharm 28, (April 1997)); where the frozen particles were in the form of large lumps of size 1-9 mm. Freezing of droplets in a moving bath of Freon 12 (-20°C), which medium conflicts with environmental demands, has been used to obtain porous, free-flowing, spherical granules with rapid dissolution (US patent 3,932,943); as well as making homogenous granules for tableting with precise dosing (US Patent 3,721,725).

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A process for preparing foamed bioabsorbable polymer particles for surgical use was presented in US Patent 5,102,983. Here, however, the porosity was very large, and the pore sizes in the range of 4 - 10 μ m, the dry content of the solution being sprayed being 1 - 20 wt%.

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US Patent 5,019,400, disclosed the use of a mixture of a biologically active material, a polymer, and a solvent which was sprayed into a non-solvent cooling medium that freezed the droplets with subsequent extraction of the solvent in the droplets during heating. The particles were finally dried in a vacuum-drier. The microparticles formed were porous, and contained 0.01 - 50 % of the active substance. The dry content of the solution sprayed was 6 wt%. This process is not entirely satisfactory since it is an advantage to have one single drying step after freezing and also a higher active substance content than 50 wt% in order to make high dose materials.

US Patent 5,405,616 discloses a method of forming droplets by forcing a suspension/solution/emulsion through calibrated jets. The droplets then fall into liquid nitrogen. Due to low shear forces the size of the pellets formed is large; 0.2 - 12 mm, which would then give a less safe dosability than if smaller particles could have been achieved. The smallest particles achieved were 0.8 - 1 mm. Further, to achieve low friability pellets, the drying step after freeze-drying was performed by thawing the pellets before conventional vacuum drying. To achieve these low friability pellets the matrix former is restricted to materials that during thawing will form a gel. The particles obtained contain no more than 33 wt% of the active substance.

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To the skilled person particle production utilizing the technique described in US 5,405,616 appears to be quite a slow process and not suitable for large scale industrial pharmaceutical production.

Object of the invention

An object of the present invention is to provide a method for the production of microparticles. More specifically, the method is for the production of homogeneous microparticles which does not have the drawbacks of the methods discussed above, e.g. methods that rely on heat or multiple solvents for drug dissolution, but instead puts no restrictions on the drug to be incorporated. A further object is to provide a method for the production of microparticles with controllable amounts of incorporated drug in a high-yield process. Also, the invention provides a method to produce homogeneous microparticles with an incorporated drug that have low friability so that they for instance can withstand coating processes. A further object of the invention is to provide a method to produce microparticles that have easily controllable density and strength. A further object is to obtain microparticles with a high content of active substance.

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Disclosure of the invention

It has now been found that free-flowing, homogeneous microparticles having low friability can be obtained by spray-freezing a suspension, solution or emulsion of a pharmaceutically active substance with subsequent freeze-drying of the frozen microparticles. The microparticles are preferably spherical in shape. The porosity of the microparticles obtained is controlled in the process by the dry content of the suspension, solution or emulsion. Apart from the porosity, the brittleness of the microparticles is controlled by the amount of polymer binder included in the suspension, solution or emulsion. In order to obtain low friability particles the dry content of the suspension or solution or emulsion should be high.

Generally the following conditions are applicable to obtain low friability microparticles according to the method of the invention;

Low friability microparticles, that can for instance withstand coating with a polymeric film, are achieved when the suspension, solution or emulsion has a dry volume content of at least 15 vol%, preferably up to 60 vol %, and a polymer binder content of at least 5 weight %, preferably 10 weight % or more, and more preferably 15 weight % or more (based upon dry content). A high total pharmaceutically active substance content can be obtained by using the present invention, such as up to 95 weight % or preferably 90 weight % (based upon dry content). The median pore size of the microparticles obtained being preferably a maximum of 1.0 µm. Dry content and dry volume content are weight % and volume %, respectively, of dry material in the suspension/solution/emulsion (dry/(dry + liquid)), wherein the dry material is pharmaceutically active substance + polymer.

According to the present invention homogenous low friability microparticles can be obtained when the dry content is from 15 to 60 vol % and the polymer binder content is 5 weight % or more giving dry microparticles with a relative density of 15 to 60 % (a

porosity of 85 down to 40 vol %). [Relative density: weight of freeze-dried material/volume of freeze-dried material/theoretical density of dry material].

The content of the pharmaceutically active substance calculated on the weight of the dried microparticles may be from 60 to 95 weight %, preferably from 75 to 90 weight %.

The dry content of the liquid medium is defined as the residue after drying at 110°C for 2 hours, divided by the total amount before drying. The dry content can be expressed either as weight percent or, preferably, as volume percent.

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The success in obtaining low porous microparticles and thus low friable microparticles depends on the volume fraction of dry material and the amount of polymer binder. The dry content of a suspension/solution/emulsion should thus preferably be expressed as a volume fraction although this cannot always be calculated.

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The microparticles may be obtained by spraying a homogeneous suspension, solution or emulsion of the active subtance(s) through an atomizer into a vessel with a cold medium with a temperature well below that of the freezing point of the liquid in the droplets. Frozen droplets will then form instantaneously. The structure of the suspension, solution or emulsion is retained in the droplets providing a homogeneous distribution of the substances within the droplets. The frozen liquid is then sublimated by freeze-drying of the frozen droplets where the structure of the droplets is retained due to lack of migration of substances during drying.

The following general steps of the procedure are further exemplified in the Experimental Section below:

a) Preparation of a medium for atomizing. The medium is a suspension, a solution or an emulsion of the active substance. A suspension may be prepared by dissolving or dispersing a polymer in a liquid (as defined below), and then adding fine particles of the

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active substance. A further dispersing agent (typically in an amount of less than 20 % (w/w) of the polymer amount) might also be included to facilitate the dispersion of the active substance. The polymer might then act as a binder between the fine active substance particles in the microparticles and can be either a water soluble or a non-water soluble polymer, according to definitions below.

- b) Atomizing of the suspension/solution/emulsion into droplets. The suspension, solution or emulsion is fed by e.g. a peristaltic pump through a nozzle that could be a pneumatic nozzle, an ultrasonic nozzle, a rotary atomizer or a pressurized nozzle. A typical size distribution of spheres produced by this process can range from 1000 µm down to 10 µm.
- c) Freezing of the formed droplets: The atomizer is situated above the cold medium in a cvlindrical vessel. If the cold medium is a liquified gas the droplets in the spray formed by the nozzle hit the cold boiling gas before hitting the cold medium that is stirred to get a better wetting of the droplets. Instant freezing takes place and the structure of the homogeneous suspension is retained within the frozen microparticles.
- d) Sublimation of the frozen liquid within the droplets: The frozen droplets are transferred from the cold medium to a freeze-drier to sublimate the frozen liquid. This step takes place without any shrinkage of the droplets or migration of excipients (e g polymers) and thus the structure of the suspension/solution/emulsion is retained within the dry microparticles.

The polymer or dispersing agent used for the formulation may be a dry polymer that is partly or fully soluble in the liquid. The polymer or dispersing agent used might also be a dispersion of polymer particles (e g a latex).

The polymer or dispersing agent could be but are not limited to the excipients listed below.

- cellulose derivatives, like ethylcellulose, hydroxypropyl methyl cellulose, hydroxyethyl cellulose, hydroxypropyl cellulose, ethyl hydroxyethyl cellulose, carboxymethyl cellulose. cellulose acetate butyrate, cellulose acetate phtalate, methylcellulose, etc
- other polysaccharides, like alginate; xanthan; carrageenan; scleroglucan; pullulan; dextran; hyaluronic acid; chitin; chitosan; starch; etc
 - other natural polymers, like proteins (e g albumin, gelatin, etc); natural rubber; gum arabic; etc
- synthetic polymers, like acrylates (e g polymethacrylate, poly(hydroxy ethyl methacrylate), poly(methyl methacrylate), poly(hydroxy ethyl methacrylate co methyl methacrylate), Carbopol® 934, etc); polyamides (e g polyacrylamide, poly(methylene bisacrylamide), etc); polyanhydrides (e g poly(bis carboxyphenoxy)methane, etc); PEO-PPO block-co-polymers (e g poloxamers, etc); polyvinyl chloride; polyvinyl pyrrolidone; polyvinyl acetate; polyvinyl alcohol; polyethylene, polyethylene glycols and co-polymers thereof; polyethylene oxides and co-polymers thereof; polypropylene and co-polymers thereof; polystyrene; polyesters (e.g. poly(lactic acid), poly(glycolic acid), poly(caprolactone), etc, and co-polymers therof, and poly(ortho esters), and co-polymers thereof); polycarbonate; cellophane; silicones (e.g. poly (dimethylsiloxane), etc); polyurethanes; synthetic rubbers (e.g. styrene butadiene rubber, isopropene rubber, etc); etc
- surfactants, e.g. anionic, like sulphated fatty alcohols (e g sodium dodecyl sulphate),
 sulphated polyoxyethylated alcohols or sulphated oils, etc; cationic, like quaternary
 ammonium and pyridinium cationic surfactants, etc; non-ionic, like polysorbates (e.g.
 Tween), sorbitan esters (e.g. Span), polyoxyethylated linear fatty alcohols (e.g. Brij),
 polyoxyethylated castor oil (e g Cremophor), polyoxyethylated stearic acid (e g Myrj), etc.

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- other substances, like shellacs; waxes (e.g. carnauba wax, beeswax, glycowax, castor wax, etc); nylon; stearates (e.g. glycerol palmitostearate, glyceryl monostearate, glyceryl tristearate, stearyl alcohol, etc); lipids (e g glycerides, phospholipids, etc); paraffin; lignosulphonates; etc.

Also, combinations of these excipients are possible.

The excipients mentioned above can be toughened by introducing a plasticizer. The plasticizer can be but is not limited to the plasticizers mentioned below.

- glycerin, polyethylene glycol, propylene glycol, triethyl citrate, diethyl phthalate, dibutyl phthalate, dibutyl sebacate, sorbitol, triacetin, etc

Also, combinations of these plasticizers are possible.

The liquid used for the preparation of the suspension/solution/emulsion, can be a solvent for the excipients listed above and encompass, e g water or organic solvents with freezing points well above the freezing point of the medium used for freezing as exemplified below. Liquids, alone or a mixture of, suitable to make a suspension/solution/emulsion of the active substance, can then be, but are not limited to:

- water (melting point (mp) 0°C), tertiary butyl alcohol (mp 25.5 °C), cyclohexane (mp +6°C), methylene chloride (mp -95.1 °C), acetone (mp -95.3 °C), methanol (mp -94 °C), ethanol (mp -117 °C), etc;
- The cold medium can typically be a liquified gas, e.g. liquid nitrogen (boiling point 196°C), liquid argon (boiling point -186°C), liquid oxygen (boiling point -183°C), or a cooled solvent well below the freezing point of the liquid in the suspension.

The mechanical strength of the microparticles is important for determining whether they will withstand processing with a polymer coating in a fluid bed.

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Examining the microparticles with a microscope before and after the fluidization in a fluid bed will give an indication of their mechanical strength.

- To achieve a relative measurement of mechanical strength the pressure where microparticles started to deform was evaluated. Microparticles within a certain size range (sieve fraction) were placed as a monolayer onto the surface of a probe with a certain area. Different loads (forces) were applied to the layer of microparticles for one minute.
- Examination of the monolayer of microparticles before and after loading was made in a Scanning Electron Microscope to see at what load the microparticles started to deform. The pressure at which the microparticles started to deform was then calculated.
 - Pharmaceutically active substances suitable to form microparticles of this invention can be but are not limited to peptides, proteins, low molecular organic substances, pro-drugs, antigens, hormones.

Thus, a microparticle according to the present invention comprises one (or several) pharmaceutically active substances with one or several additional non-active substances, which are dispersed within the microsphere.

Uncoated particles can be retrieved as they are easily dissolved when they are immersed into a liquid due to their porous structure.

The microparticles obtained can be coated with a polymer to achieve either a timecontrolled release, a site-controlled release or a pH-dependent release. Suitable polymers for coating can be, but are not limited to, the same type of polymers as listed above.

The coated microparticles can be put into capsules or incorporated into a tablet compressed by methods known by those skilled in the art.

The formulations produced based on the microparticles, coated or uncoated, can be given by different administration routes, such as, but not limited to, the oral, the parenteral, the nasal, the pulmonary, the rectal, the tonsillar, the buccal, the intraocular, the vaginal etc, administration routes. The preferred administrations are by the oral, nasal, pulmonary and rectal routes.

Working examples

10 The following examples illustrate different aspects of the invention.

The size distribution of the obtained microparticles was measured by sieving. By mercury porosimetry measurements the bulk-density and pore-size distributions were determined. To determine the median pore size the pressure range for mercury intrusion corresponded to pore sizes between $0.0005 \mu m$ and $10 \mu m$.

By subjecting a monolayer of the microparticles to compaction forces their relative strength was measured.

EXAMPLE 1: Preparation of microparticles with a high loading of dry content that withstand coating in a fluidized bed

A suspension containing talc powder was made according to the composition below;

Talc powder (1-2 μm)	90 g
Talc powder (5 - 80 μm)	210 g
HPMC, 6 cps	80 g
Tween 80 (polysorbate 80)	6 g
Purified water	750 g

Weight percent of dry content in suspension: 34 (19.2 vol%)

First, polysorbate 80 was mixed with the water. The HPMC was then added and dissolved during stirring with subsequent addition of the substance. The suspension was then deagglomerated by high-shear mixing. The deagglomerated suspension was fed through a pneumatic nozzle with a diameter of 1.0 mm at a speed of about 15 ml/min. The pressure of the atomizer was 1 bar. The spray formed first hit the cold gas above the liquid in a vessel filled with liquid nitrogen that was stirred to get a better wetting and instantaneous freezing of the droplets. The frozen droplets have a higher density than liquid nitrogen which make them sink to the bottom of the vessel. The frozen droplets/microparticles were then placed in a conventional freeze-drier with a shelf-temperature of -20°C. The primary drying was performed stepwise at -20°C to 0°C at 0.1 mbar. The dry microparticles were free-flowing and spherical. Scanning Electron Microscopy showed a homogeneous distribution of the talc powder with pores (0.1 - 2 μm) in between. The bulk density, median pore size and mechanical strength was measured and the results are shown in table 2.

Compaction measurements showed that the microparticles obtained had a low friability (high mechanical strength).

Fluidization of the microparticles in Example 1 in a fluidized bed showed by microscopy that the microparticles did not break down. These microparticles started to deform at a pressure of 94 kPa (sieve fraction: $450-630 \mu m$). Final coating with a polymer in a fluidized bed proved that the microparticles could be successfully coated.

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EXAMPLE 2: Coating of microparticles with a polymeric film

The microparticles from Example 1 were easily handled without falling apart and tough enough to be successfully coated. A fraction of 20 g of the microspheres, $150 - 300 \, \mu m$ in

size, were successfully coated with an enteric polymer to a film thickness of 30 $\mu m,\,in$ a fluidized bed.

Characterization of pellets obtained in example 1.

Table 1. Size distribution. Sieving (weight fraction%)

Fraction	Example 1
< 100 μm	1
100 - 150 µm	2
150 - 300 μm	22
300 - 450 μm	32
450 - 630 μm	26
630 - 800 μm	12
800 - 1000 μm	4

Table 2. Characterization of microparticles

Example no.	Dry content (vol%)	Binder (wt%) based on dry	Mercury poro	·	Mecha	nical strength
		content	Bulk density (g/cm³)	Pore median size (µm) (measured range: 0.0005- 10 µm)	Kpa	Fraction
1	19.2	21	0.47	0.8	94	450 - 630 μm

CLAIMS

- 1. A method of preparing homogeneous microparticles containing a pharmaceutically active substance by use of a spray freezing technique which method comprises
- atomizing into droplets a liquid medium having a minimum dry content of 15% by volume and comprising
 - a) a pharmaceutically active substance,
 - b) a polymer selected from the group consisting of water soluble polymers and non-water soluble polymers, said polymer being present in an amount of at least
 5 per cent by weight based upon the dry content of the medium,
 - c) a liquid in which the pharmaceutically active substance and polymer are suspended, dissolved or emulsified, and
 - d) optionally a dispersing agent, selected from the group consisting of polymers, surfactants, other substances and mixtures thereof,
- freezing the formed droplets and
 - sublimating the frozen liquid of the droplets to obtain dry, homogeneous microparticles.
- 2. A method according to claim 1, wherein the polymer of the liquid medium constitutes
 10 weight % or more of the dry content.
 - 3. A method according to claim 1, wherein the polymer of the liquid medium constitutes 15 weight % or more of the dry content.
- 4. A method according to claim 1 wherein the dry content of the liquid medium is from 15 to 60 vol %.
 - 5. A method according to claim 1, wherein the dry volume content of the liquid medium is from 15 to 60 vol % and gives dry microparticles with a relative density of 15 to 60 %.

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- 6. A method according to claim 1, wherein the dry volume content of the liquid medium is from 15 to 60 vol % and gives dry microparticles with a porosity of 85 down to 40 vol %.
- 5 7. A method according to claim 1 wherein the liquid medium to be spray-freezed is a suspension.
 - 8. A method according to claim 1 wherein the liquid medium to be spray-freezed is a solution.
 - 9. A method according to claim 1 wherein the liquid medium to be spray-freezed is an emulsion.
- 10. A method according to any of the preceding claims wherein the content of the
 pharmaceutically active substance is from 60 to 95 weight %, preferably 75 to 90 weight
 %, of the weight of the dried microparticles.
 - 11. A method according to any of the preceding claims wherein the dry content of the medium is from 15 to 60 vol% and with the content of the pharmaceutically active substance being from 60 to 95 weight % of the dried microparticles.
 - 12. A method according to any of the preceding claims wherein the polymer is selected from the group consisting of a cellulose derivative, a polysaccharide, a natural polymer, a synthetic polymer, a surfactant and mixtures thereof.
 - 13. A method according to any of the preceding claims wherein the dispersing agent is selected from the group consisting of polymers, surfactants, other substances and mixtures thereof.

- 14. A method according to any of the preceding claims wherein the liquid in which the polymer is soluble is selected from the group consisting of water, tertiary butyl alcohol, cyclohexane, methylene chloride, methanol, ethanol and mixtures thereof.
- 15. A method according to any of the preceding claims wherein the cold medium is selected from the group consisting of liquid nitrogen, liquid argon, liquid oxygen or a cooled solvent well below the freezing point of the liquid in the suspension.
- 16. A method according to any of the preceding claims wherein the sublimation isperformed by freeze-drying.
 - 17. A method according to any of the preceding claims wherein the size distribution of the prepared microparticles are in the range from 10 to 1000 µm.
- 18. Microparticles when prepared according to the method of any of claims 1-17.
 - 19. The microparticles according to claim 18 further comprising a polymeric film coating.
- 20. A method of preparing homogenous microparticles containing a pharmaceutically active substance, the particles being coated with a polymer film coating, which method comprises a method as claimed in any one of claims 1-17 followed by coating the microparticles with a polymeric film coating.

INTERNATIONAL SEARCH REPORT

Box 5055, S-102 42 STOCKHOLM

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International application No.

PCT/SE 00/01682 A. CLASSIFICATION OF SUBJECT MATTER IPC7: A61K 9/14, A61K 9/50 According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC7: A61K Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched SE,DK,FI,NO classes as above Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) C. DOCUMENTS CONSIDERED TO BE RELEVANT Category * Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. A US 5102983 A (JOHN KENNEDY), 7 April 1992 1-20 (07.04.92)GB 2329124 A (HIRAN ASOKA MALINGA RATWATTE), A 1-20 17 March 1999 (17.03.99) A US 5017383 A (YASUO OZAWA ET AL), 21 May 1991 1-20 (21.05.91)A WO 9013285 A1 (ENZYTECH, INC.), 15 November 1990 1-20 (15.11.90)Further documents are listed in the continuation of Box C. See patent family annex. Special categories of cited documents: later document published after the international filing date or priority document defining the general state of the art which is not considered date and not in conflict with the application but cited to understand to be of particular relevance the principle or theory underlying the invention earlier application or patent but published on or after the international "X" document of particular relevance: the claimed invention cannot be filing date considered novel or cannot be considered to involve an inventive "L" document which may throw doubts on priority claim(s) or which is step when the document is taken alone cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination document referring to an oral disclosure, use, exhibition or other being obvious to a person skilled in the art document published prior to the international filing date but later than "&" document member of the same patent family the priority date claimed Date of the actual completion of the international search Date of mailing of the international search report 0 9 -01- 2001 <u>2 January 2001</u> Name and mailing address of the ISA/ Authorized fficer Swedish Patent Office

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